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The RIAR **DOVITA-1/2** P&T Program – Current Results of the 15-year R&D activities

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Experience in Closed Fuel Cycle including pyrochemical processes



Research Institute on Atomic reactors – RIAR (Dimitrovgrad) is most Russian R&D center of non-aqueous methods development:

- ◆ Pyrochemical investigation - from early 1960-s
- ◆ Demonstration of fluoride volatility reprocessing technology – 1970s
- ◆ Demonstration facility for pyrochemical MOX-fuel production for fast reactor – from late 1970-s
- ◆ Pyroelectrochemical reprocessing experience – 1990-2003
- ◆ Preparation for industrial application – from 1990-s
- ◆ Study on transmutation cycle, nitride fuel and other applications – from 1990s

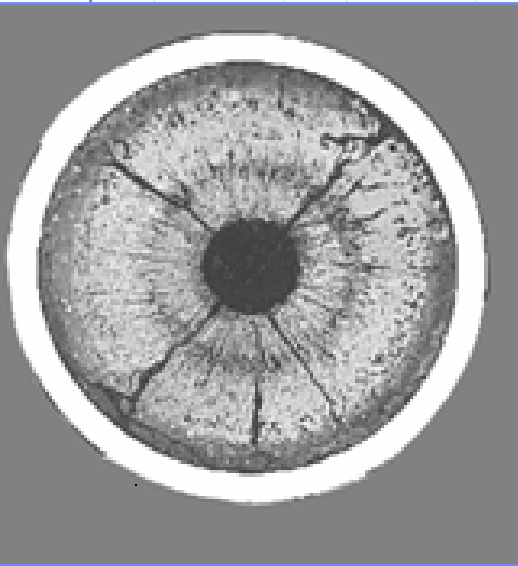


Since 1992 RIAR has been performing own R&D **DOVITA** Program



- **D**ry technologies for MA fuel reprocessing and preparation
- **O**xide fuel application as the most widely studied one
- **V**ibropacking automated technology of the fuel pin production
- **I**ntegrated disposition of fuel reprocessing and fuel element refabrication facilities on the same site with the reactor
- **TA** The whole complex of approaches will permit a creation of the compact plant for Transmutation of Actinides

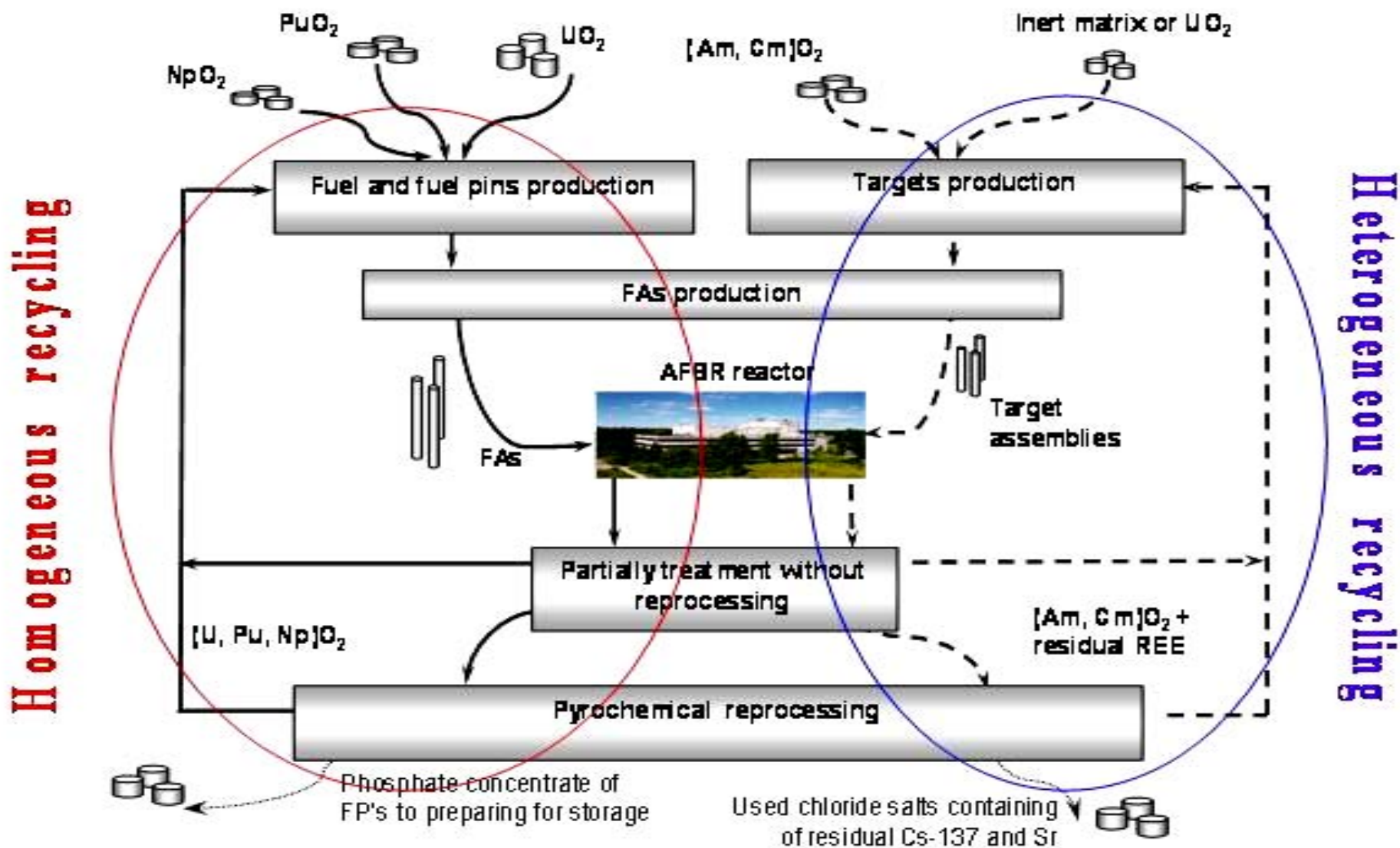
Experience in adding of minor actinides in fuel compositions and results of irradiation



Irradiated
(U,Np)O₂ fuel,
19% burn-up

- ✓ Pyrochemical technology of adding Np into oxide fuel (5-20%) has been developed
- ✓ Performance of vi-pack fuel with (U,Np)O₂ fuel has been validated experimentally to ~20% burnup in BOR-60
- ✓ No evidence of significant difference in performance of fuel rods with (U,Np)O₂ fuel compared with UO₂ or MOX fuel rods has been noticed
- ✓ Pyrochemical process of codeposition of Am with MOX fuel (2-4%) has been developed
- ✓ Methods of Am/REE separation in melts has been tested
- ✓ Special vi-pack targets containing Am oxide with UO₂ or inert matrix have been developed
- ✓ Transmutation of Np, Am, Cm is being studied in BOR-60

DOVITA fuel cycle



New times consideration:

DOVITA



DOVITA-2



1992

- **D**ry technologies
- **O**xide fuel with MA
- **V**i-pack
- **I**ntegrated disposition same site with the reactor
- **TA** Transmutation of Actinides

2006-2008

- ❖ **D**ry technologies
- ❖ **O**n-site reprocessing
- ❖ **V**arious type of fuel with MA
- ❖ **I**ntegration of MA recycling into FR Closed Fuel Cycle
- ❖ **TA** - Transmutation of Actinides

DOVITA-2



Fuel type/ Stages	Oxide vi-pack	Oxide pellet	Nitride pellet/ vi-pack	Metal	Molten salt
Concept Studies	+	+	+	+/-	+
R&D	+	-/+	+/-	-	+
Fuel Production	+	-	-	-	-
Irradiation Testing	+	-	-	-	-
PIE	+	-	-	-	----
Reprocessing	-/+	-	-/+	-	+/-

DOVITA-1

Research area of pyrochemical and non-aqueous MA/FP partitioning in frame DOVITA-2 Program



Melts/ media	Electrochemical methods				Oxides Precipitation	Metallization
	Liquid cathodes	Solid cathodes	Anodic dissol-n	E/Chem. oxide titration		
Chlorides	Cd, Bi, Ga, Al	Al	+	+	+	+
Fluorides	Bi, Ga, Al, Pb	-	+	+	+	-
Cl ⁻ /F ⁻	Cd, Bi, Ga, Al, Pb	Al	+	+	+	-
Molibdates	-	+	-	-	+	-
RTIL	Ga	Zn, Cu, Ni, Al, ...	+	-	-	-

+ Methods of Volatility of Fluorides and /or Chlorides

Recent RIAR activities in frame of DOVITA-2 Program

ISTC Project # 3261 (2006-2008)

Study of Cm thermodynamics in molten chlorides



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Objectives

Reactions of formation for oxygen Cm compounds

- equilibrium constants for reactions of Cm oxygen compounds formation versus temperature;
- equilibrium constants for reactions of Cm oxygen compounds formation versus the inverse effective radius of solvent cation

1st Year is completed

Reactions of formation for oxygen-free Cm compounds

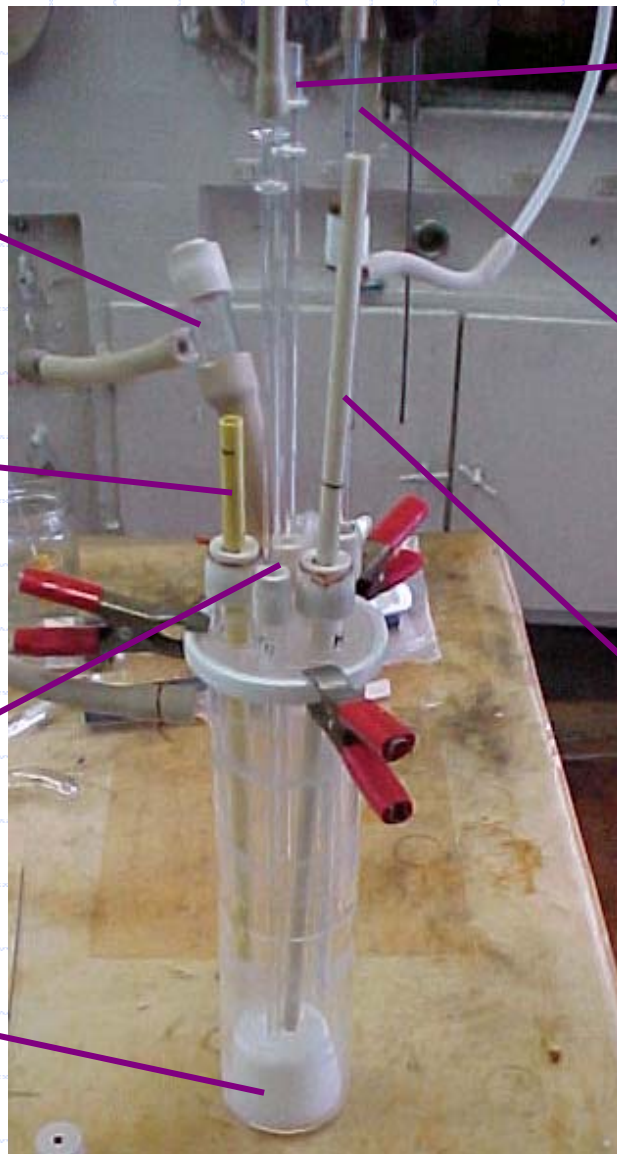
- standard potential for redox pair versus temperature;
- standard potential for redox pair versus the inverse effective radius of solvent cation

Simulation of Cm behavior in molten chlorides

Pourbaix diagrams

Collaborators - ITU, CIEMAT, CEA, KTH

ISTC project #3261. Current Results



Sluce

Thermocouple

Oxygen sensor

Reference electrode Ag/Ag⁺

Gas supply tube

Oxygen pump

Salt

Potentiometric titration of CmCl₃



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ISTC project #3261. Current Results

Potentiometric titration of CmCl_3



Applied methods:

- **main - titration by BaO**
- **combined - in case of absence of Section on preliminary experimental curve for CmO^+ formation :
titration by current through the electric circuit
“oxygen pump” – “counter electrode” and then by BaO**



ISTC project #3261. Current Results



Oxygen pump
 $\text{ZrO}_2(\text{Sc}_2\text{O}_3) \mid \text{Pt}(\text{O}_2)$

Reference electrode
 $\text{Pyrex} \mid \text{Salt} + \text{AgCl}(0.75\text{mol/kg}) \mid \text{Ag} \mid \text{Pt}(\text{Ar})$

Cm dioxide deposit

Counter electrode
 $\text{Pyrex} \mid \text{Salt} \mid \text{C}$

Oxygen sensor
 $\text{ZrO}_2(\text{Y}_2\text{O}_3) \mid \text{Pt}(\text{O}_2)$

Oxygen pump after experiment



INIAP

ISTC project #3261. Current Results

Conditions of experiments



Salt	Amount of Runs	Temperature range, °C	Range of Cm content in melt, mole/kg
NaCl-2CsCl	11	550-750	$(1.1-14) \cdot 10^{-3}$
3LiCl-2KCl	7	450-650	$(1.9-7.9) \cdot 10^{-3}$
NaCl-KCl	3	750-850	$(5.0-6.9) \cdot 10^{-3}$

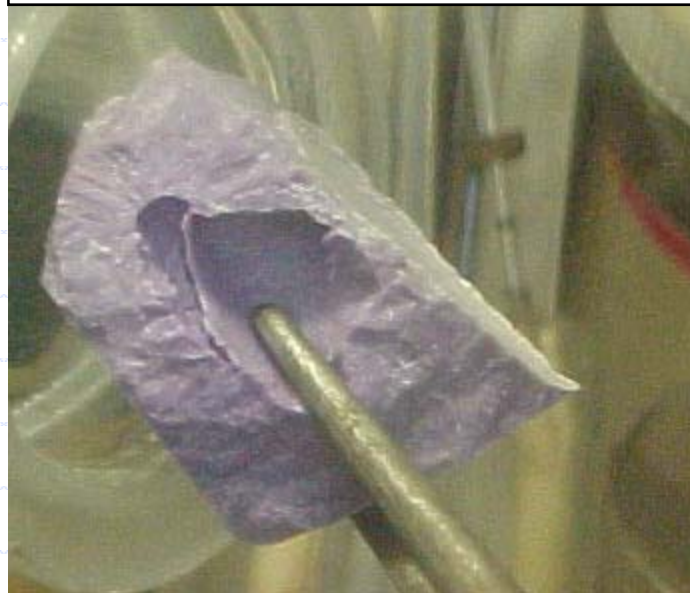


ISTC project #3261. Current Results

$\text{NaCl-2CsCl+CmCl}_3(1.4 \cdot 10^{-2} \text{mole/kg})$



$\text{LiCl-KCl+CmCl}_3(5.5 \cdot 10^{-3} \text{mole/kg})$



$\text{NaCl-KCl + CmCl}_3(5.0 \cdot 10^{-3} \text{mole/kg})$



Salt ingot with CmCl_3



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ISTC project #3261. Current Results

Experimental constants of CmO^+ , CmOCl and Cm_2O_3 dissociation in molten chlorides at different temperatures, molality scale



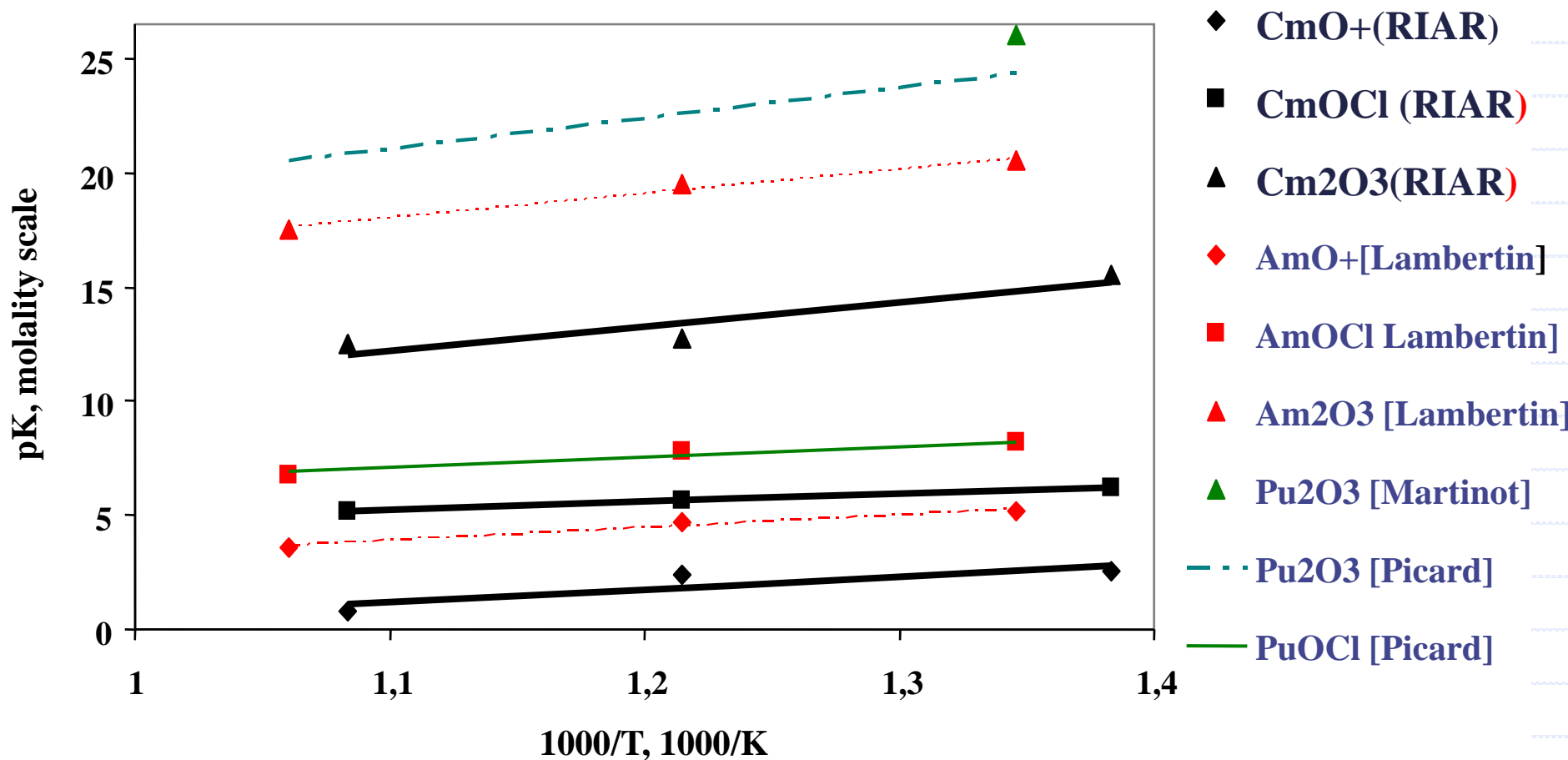
Melt	T, °C	pK(CmO^+)	pK(CmOCl)	pK(Cm_2O_3)
NaCl-2CsCl	556	4.2 \pm 0.2	7.9 \pm 0.2	20.1 \pm 0.3
	650	3.4 \pm 0.2	7.5 \pm 0.2	18.5 \pm 0.3
	750	3.7 \pm 0.2	6.7 \pm 0.2	16.8 \pm 0.3
3LiCl-2KCl	450	2.5 \pm 0.2	6.2 \pm 0.2	15.5 \pm 0.5
	550	2.4 \pm 0.2	5.7 \pm 0.2	12.7 \pm 0.5
	650	0.83 \pm 0.1	5.2 \pm 0.2	12.5 \pm 0.5
NaCl-KCl	750	2.6 \pm 0.2	5.9 \pm 0.2	12.9 \pm 0.4
	800	2.4 \pm 0.2	5.8 \pm 0.2	12.6 \pm 0.4
	850	1.3 \pm 0.1	5.6 \pm 0.2	12.1 \pm 0.4



ISTC project #3261. Current Results

Comparison with literature data on other actinides.

Dependence of dissociation constant for actinides upon temperature in LiCl-KCl



Recent RIAR activities in frame of DOVITA-2 Program



- ◆ R&D for production of vi-pack oxide Fuel with MA
 - Under continuation
- ◆ R&D for production of vi-pack nitride Fuel with MA
 - Under development
- ◆ R&D for production oxide pellet fuel with MA
 - Under development
- ◆ R&D for production metallic fuel with MA
 - Feasibility Stage
- ◆ Molten Salt chemistry of Am and Cm for fluorides
 - Under preparation
- ◆ Partitioning of MA/REE in molten chlorides/fluorides
 - liquid cathodes
 - reductive extraction
 - Oxide precipitation

Official Investment Frames for Russian Nuclear Renaissance



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- ◆ **Federal Tasks Program “Development of Nuclear Power Complex of Russia on a period of 2007 - 2015” - accepted in 2006**
- *NPP construction*
- ◆ **Federal Tasks Program “Nuclear and Radiation Safety” (2008-2015) - accepted in 2007**
- *RAW Heritage*
- ◆ **Federal Tasks Program “New Generation Nuclear Energy Technologies” (2010-2020) – on a final preparation Stage**
- *Innovations*

Federal Tasks Program

“New Generation Nuclear Energy Technologies”

RIAR planned participation



- **Multi-functional Fast Research Reactor (MFRR) - 2016**
- **Large Multi-Purpose Pyrochemical Reprocessing Complex - 2015**
 - **Molten salt Reprocessing Facility**
 - ✓ capacity – up to 2 500 kg of SNF per Year (fuel type: oxide, nitride, metallic, IMF)
 - **Fluoride volatility Reprocessing Facility,**
 - ✓ capacity – up to 500 kg of SNF per Year (mainly – LWR SNF)
- **New Lab for Experimental and Innovative Fuel Production – 2010-1012 (incl. Fuel and Targets with MA)**
- **Demonstration of Closing Fuel Cycle based on Pyrochemical technologies -2016-2020-... on a levels:**
 - **Up to 50 spent FAs of BN-600/800**
 - **Full scale CFC for MFRR from initial fuel loading**
 - **Other experimental implementations**

New Russian Sodium Fast Research Reactor – Multi-functional Fast Research Reactor (MFRR)

Location – RIAR site



Characteristic	Value
Maximum flux Φ_{\max} , n/cm ² ·sec	~ 6.0·10 ¹⁵
Thermal power, MWth	~ 150
Electric power, MWe	~ 50
Number of independent experimental loops (~1 MWth, sodium, heavy metal and gas coolant + salt coolants)	3 (+1 behind reactor vessel)
Driven Fuel	Vi-pack MOX, (PuN+UN)
Core height, mm	400-500
Maximum heat rate, kW/l	1100
Fuel Cycle	Full Scale Closed FC based on Pyro Processes
Test Fuel	Innovative Fuels, MA Fuels and targets
Maximum fluence in one year, n/cm ²	~ 1,2·10 ²³ (up to 55dpa)
Design lifetime	50 year
RR creation time (no more than, years)	9 (2008 – 2016)

Start of BN Closed fuel Cycle based on RIAR technologies



- ◆ **2011** - start of vi-pack MOX-fuel production for BN-800
- ◆ **2012** – start of BN-800 operation
- ◆ **2016...2018** – demonstration of BN-800 closed fuel cycle technologies

Key final official decisions:

- ◆ MOX-fuel production by pyroelectrochemistry and vibropacking
- ◆ Trend to closing of fuel cycle by compact dry technologies
- ◆ Development and testing of new fuel and new technologies

RIAR R&D International cooperation in the field of advanced FC

	Fuel production		Repro- cessing	P&T	Other	Cladding material s	Concept Studies	Funda- mental Studies
	MOX	other						
France	-	MA oxide	-	Am/Cm recovery	Pyro	+	FS	Cm
INPRO	-	-	-	-	-	-	CPP RUS-2	-
Japan	MOX vibro	-	MOX	MA/REE separ.	Fluorex/ MoO ₄ ²⁻	ODS	FS	MA
Korea	-	-	Metalliz. / vibro- DUPIC	MA/REE separ.	Pyro	-	-	-
US?	TRU fuel ?		UREX+1 ?	TRU fuel?	-	-	-	<i>Pu in RTIL's</i>
EU	-	MA nitride	-	-	MSR fuel	-	-	Cm

Red color – DOVITA-1/2 activities